## **Chemical Engineering 140** Chemical Process Analysis Spring 2019

Instructors:	C.J. Radke	101E Gilman, radke@berkeley.edu Office Hour: Tues 3-4 p.
Teaching Assistants:	Sarah Berlinger	sarah_berlinger@berkeley.edu, Office Hour: Wed 5-6 p, 100E Chem Lib
	Further consultation with either the instructor or the teaching assistants is available by individual appointment.	
Objectives:	CBE 140 introduces the principles of mass and energy balances along with equilibrium and rate expressions. Application of these principles is made towards the solution of basic chemical engineering processing problems. This course is foundational for the chemical engineering curriculum.	
Text:	Required:Felder and Roussearand Sons, Inc. 3rd ERecommended (on 2 hr reR. M. Murphy, IntroChem. Lib. Call NuHimmelblau and RiEngineering. Prenti2012Duncan and ReimerIntroduction. CambTP155.D74 1998Russell and Denn.and Sons, Inc. 1972R. N. Shreve, The OKresge Engineering	<ul> <li>a. Elementary Principles of Chemical Processes. John Wiley Edition, 2005.</li> <li>serve in the Chemistry or Engineering Library): oduction to Chemical Processes, McGraw Hill, 2007</li> <li>amber: TP155.7.M87 2007</li> <li>ggs. Basic Principles and Calculations in Chemical ice-Hall, 8<sup>th</sup> Ed. 2012. Chem. Lib. Call Number: TP151.H5</li> <li>c. Chemical Engineering Design and Analysis: An ridge University Press. 1998 Chem. Lib. Call Number:</li> <li>Introduction to Chemical Engineering Analysis. John Wiley</li> <li>Chem. Lib. Call Number: TP155.R88</li> <li>Chemical Process Industries. McGraw Hill, 1956.</li> <li>g Lib. Call Number: TP145.S5 1956</li> </ul>
Description:	Analysis of chemical processes depends on the ability to construct balances on material and energy within a system. Subsequent courses in the curriculum will build on this skill by elaborating on the selection and nature of different terms in these balances. Class examples will be drawn from standard chemical engineering unit operations and processes, and some homework will be solved using spreadsheet and computational computer software. The text is only followed loosely; students are encouraged to refer to other recommended texts when necessary.	
Course Grade:	The course grade will be o	letermined by the following:

nework:	10% (lowest 2 scores will be dropped if course evaluations are turned in)
ign Report	15 %
term Exams (2):	22.5 % each (02/27, 04/10, location, time:
	TBA)
ıl Exam:	35 % (05/15)
ign Report term Exams (2): ıl Exam:	evaluations are turned in) 15 % 22.5 % each ( <b>02/27, 04/10</b> , location, time: TBA) 35 % ( <b>05/15</b> )

**Homework:** Homework will be assigned on Monday of each week and will be due online at the bourse website by midnight on Wednesday evening one week later. No late homework will be accepted. Assignments, solutions, and handouts will be posted at the class becourse website.

**Examinations:** There will be no regrades of examinations. Use of electronic devices with access to internet is not permitted.

Tutoring: https://chemistry.berkeley.edu/ugrad/current-students/tutoring