MSE 102: Bonding, Crystallography and Defects Syllabus – Fall Semester, 2020 Instructor: Professor Daryl C. Chrzan

Date	Торіс	Problem Sets
08/27	First day of class: logistics, mathematical background	
Part I:	Symmetry, Crystallography and Crystal Structures	
09/01	Introduction to lattices. Coordinates, directions, planes. BO. Ch. 2, KGK Ch. 1, R, Ch. 2, S Ch.1	
09/03	Lattices continued, introduction to symmetry operations. BO. Ch. 5, R Ch. 3, S Ch. 2	
09/8	Symmetry operations, symmetry operations compatible with lattice translations.	PS #1 Due
09/10	Improper rotations, glide planes, screw axes. 2D plane lattices.	
09/15	3D Bravais lattices; Introduction to crystal structures. BO. Ch. 3&6	PS #2 Due
09/17	Crystal structures: Lattice+basis. Stereographic projections. KGK Ch. 2	
09/21	@HOME : Quiz #1 All things lattice based, crystal definitions.	
09/22	32 crystallographic point groups. 230 space groups.	
09/24	Introduction to the International Tables for Crystallography.	PS #3 Due
09/29	Relationship between symmetry and physical properties: Reciprocal lattices and diffraction. AM reciprocal lattice, R Ch. 5	
10/01	Diffraction continued. N Ch. 1&2, KGK Ch. 4	PS #4 Due
10/05	@HOME: Quiz #2 Symmetry, Diffraction and the International Table	S
10/06	Introduction to tensors: Conductivity. General transformation properties	

10/08	Symmetry constraints on tensors.			
10/13	Linear elasticity theory, elastic constants. N Ch. 7&8, KGK Ch. 5	PS #5 Due		
10/15	Tensors continued.			
Part II:	Bonding and Crystal Binding			
10/20	Intro to bonding. Van der Waal's bonding.	PS #6 Due		
10/22	Van der Waal's bonding continued, introduction to ionic bonding.			
10/26	@HOME: Quiz #3 Tensors			
10/27	Ionic bonding continued.			
10/29	Ionic bonding: ionicity and electronegativity. R Ch. 7			
11/03	The need to think quantum mechanically. Introduction to Schrödinger's equation. Particle in a box.	PS #7 Due		
11/05	Quantum mechanical picture of bonding. Covalent and ionic limits.			
11/09	@HOME: Quiz #4 Classical Models of Bonding van der Waals and ionic bor	nding		
11/10	Solution to H atom; relationship to periodic table.			
11/12	Covalent bonding. Introduction to the band theory of solids. AM Bloch, R Ch. 8, KP-Lec.pdf (under lecture notes)			
11/17	Continued Discussion of the band theory of solids.	PS #8 Due		
Part III: Point and Line Defects				
11/19	Point defects and equilibrium. Point defects in ionic materials.			

11/23 @HOME: Quiz #5 Covalent Bonding, Band Theory of Crystals

- 11/24 Point defects in covalent and metallic materials.
- 11/26 Thanksgiving Holiday. No class.

12/01 Point defects continued. Intro to dislocations.	
12/03 Dislocations continued.	PS #9 Due
12/?? Final Exam	

B.O. = Borchardt-Ott; KGK = Kelly, Groves and Kidd; R = Rohrer; S = Sands; N = Nye; AM = Ashcroft and Mermin