Chem 112B Midterm 2

Instructor: Richmond Sarpong March 23rd 2017 8:10-9:30 am, 100 Lewis

You have **80 minutes** to complete this exam. Please write you answers clearly only on the pages indicated <u>and be as detailed as possible</u>. Nothing written outside the numbered pages will be graded. There should be 8 total pages in this exam.

Name: KEY	
UID:	
GSI Name:	
Question	Score
I	(14 points)
2	(15 points)
3	(15 points)
4	(17 points)
5	(20 points)
6	(19 points)
Total	(100)

Question 1 (14 points):

Provide reagents to accomplish the transformations shown below. Note that a given transformation may require multiple steps (2 pts each; 14 pts total):

Question 2 (15 points):

(a) Rank the following molecules (from 1 to 4) in terms of increasing oxidation level (1 for the lowest level). (4 pts total)

(b) Provide conditions for the following **deprotection** steps (i.e., removal of the protecting group). (2.5 pts each)

(c) Provide the names associated with the following cross-coupling reactions (6 pts)

Question 3 (15 points):

(a) A Swern oxidation provides aldehyde **B** from alcohol **A** whereas a Jones oxidation provides acid **C**. Provide a detailed mechanism for the formation of **B**: (9 pts)

(b) Explain with a mechanism how C forms from A under the Jones oxidation conditions. In addition, you may add up to two sentences (6pts)

Question 4 (17 Points):

(a) Provide reagents for the following transition metal-catalyzed transformations (3 pts each; 12 pts total)

(b) Rank the following benzene derivatives from 1 to 5 in terms of rate of reaction in a S_N Ar reaction with 1 being the fastest. (5pts)

Question 5 (20 points):

Provide a synthesis of A (shown below) from benzene and any other reagents you deem necessary. For each step, clearly indicate the equivalents of reagents that you need.

Question 6 (19 points):

Mixed anhydride XX (shown below) forms an acylium intermediate in the presence of a Lewis acid. (a) Circle the positions on pyrrole and on indole that will react first with the acylium intermediate that is

formed from XX. (2 pts each)

(b) Provide a mechanism for the S_EAr reaction of pyrrole and **XX** in the presence of a Lewis acid (LA) (15 pts)