CHEMISTRY 112A FALL 2016

EXAM 3

NOVEMBER 22, 2016

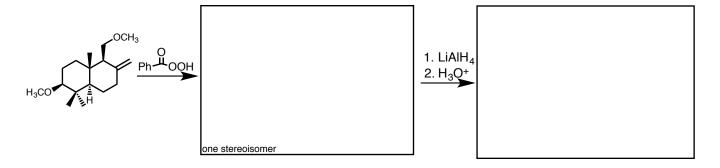
NAME- WRITE BIG _		
STUDENT ID:		
SECTION AND/OR GSI IF Y	OU ARE IN THE LABORATORY COURSE:	

- You will have 75 minutes in which to work.
- BE NEAT! Non-legible structure drawings will not be graded.
- Only answers in the answer boxes will be graded you can write in other places, but we only grade the answers in the boxes.
- All pages of the exam must be turned in.
- No calculators
- No stencils
- Molecular models may be used

Problem	Points (Maximum)
1	28
2	12
3	24
4	18
5	26
6	12
Total	120

1. (28 points) For each reaction draw the major organic products, **including all stereoisomers**. Write NR if you think there will be no reaction.

a.

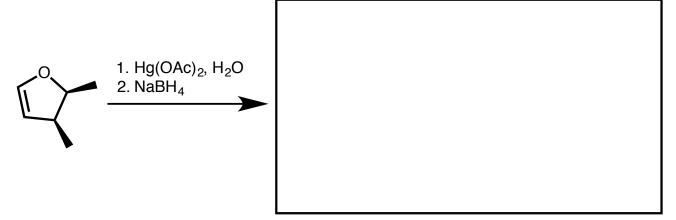


b.

$$\frac{Br_2}{CCI_4}$$

c.

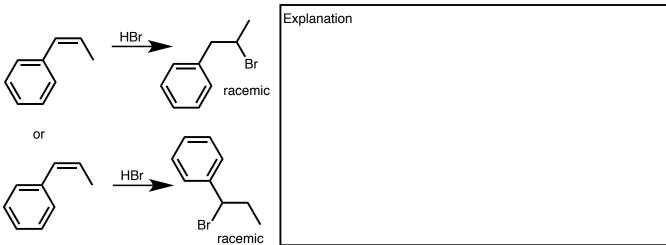
d.



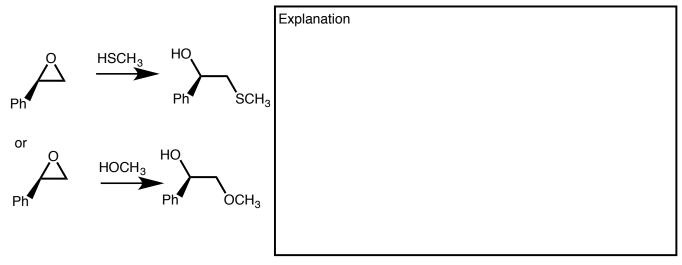
e.

2. (12 points) **Circle** the reaction in the following pairs of reactions that you would expect to go faster. It is possible that both reactions have the same rate. It is possible that one of the reactions shown in each pair does not occur at a measurable rate. You may disregard any other products besides the ones pictured that may form under the reaction conditions. Give explanations in the boxes provided.

a.



b.



3. (24 points) The following reactions would not occur as written. i. What product would actually be made? ii. Why was the desired product not formed? iii. How could you change either the substrate **or** reaction conditions to give the desired product?

a.



What product is actually made? (Draw structure or NR for no reaction)

Why was desired product not formed? How could substrate or reaction be (Explain in 1 sentence)

Constructive or NR for no reaction in 1 sentence)

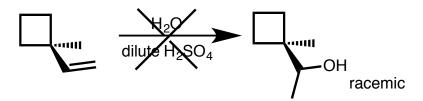
Why was desired product not formed? How could substrate or reaction be changed to give desired product?

b.



What product is actually made? (Draw structure or NR for no reaction)	Why was desired product not formed? (Explain in 1 sentence)	How could substrate or reaction be changed to give desired product?

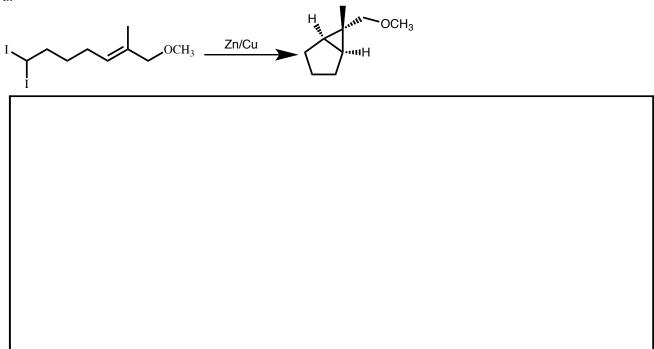
c.



What product is actually made? (Draw structure or NR for no reaction)	Why was desired product not formed? (Explain in 1 sentence)	How could substrate or reaction be changed to give desired product?

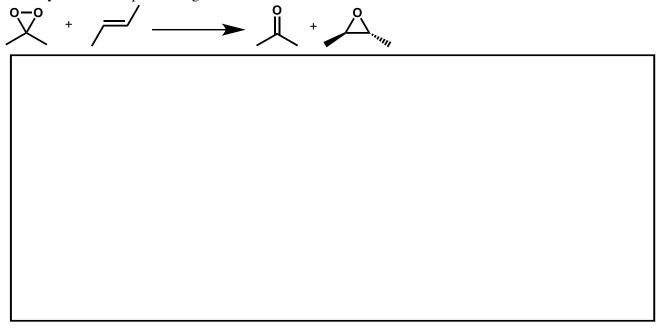
4. (18 points) Draw the mechanism of the following reactions using arrows to indicate the flow of electrons. Make sure to clearly indicate stereochemistry.

a.



b.

c. 3,3-Dimethyldioxirane is a potent epoxidation reagent. Draw the mechanism for 3,3-dimethyldioxirane epoxidizing the alkene shown below.

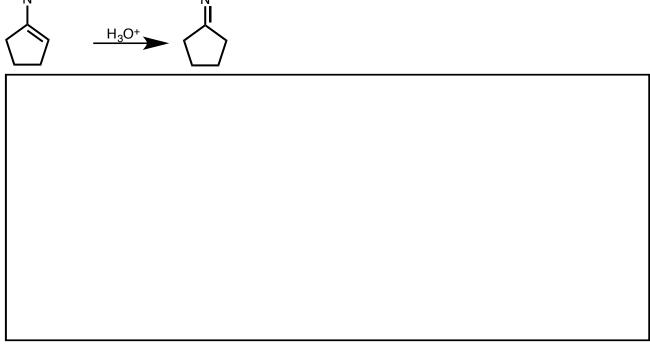


5. (26 points) Alkynes undergo hydroboration to initially form an enolate, which then tautomerizes to an aldehyde or ketone.

a. Sketch the orbitals that initially interact when HBR_2 reacts with the alkyne. Label each orbital and label it as a LUMO or HOMO.

b. Would you expect the initially formed boron addition product to be cis or trans? Explain your choice include a reaction energy diagram and a sketch of the transition state of the rate-determining step in your explanation.
e. Draw a mechanism for the base catalyzed tautomerization of the enol to the aldehyde.

d. Enamines are sim	ar to enols with nitrogen replacing oxygen. The enamine shown below will
undergo an acid-cata	yzed tautomerization to the imine. Draw the mechanism of this reaction.
H_N_CH ₃	CH₃
N 	N II
人	火



6. (12 points) Synthesize the indicated product from the indicated starting materials and any other reagents. In your synthesis, show each product formed by each set of reagents.

