CE 111 MT #1 Answers

- 1. 960 g m-3
- 2. diameter (or volume)
- 3. viscosity & density
- 4. C6H12O6 → 3 CO2 + 3 CH4
- 5. Some carbon atoms are reduced from 0 to -4 (in methane) while others are oxidized from 0 to  $\pm$ 4 (in CO2)
- 6. 51%
- 7. Material balance and Henry's law
- 8. pH will decrease. Creation of new CO2 leads to formation of carbonic acid, some of which dissociates, releasing H+.
- 9. 1.4 atm