CE 111
MT \#1
Answers

1. $960 \mathrm{~g} \mathrm{~m}-3$
2. diameter (or volume)
3. viscosity \& density
4. $\mathrm{C} 6 \mathrm{H} 12 \mathrm{O} 6 \rightarrow 3 \mathrm{CO} 2+3 \mathrm{CH} 4$
5. Some carbon atoms are reduced from 0 to -4 (in methane) while others are oxidized from 0 to +4 (in CO2)
6. $51 \%$
7. Material balance and Henry's law
8. pH will decrease. Creation of new CO 2 leads to formation of carbonic acid, some of which dissociates, releasing $\mathrm{H}+$.
9.1 .4 atm
